# A STUDY ON MAGNESIUM, COPPER AND ZINC METABOLISM IN SERUM AND URINE OF HYPOTHYROID PATIENTS BEFORE AND AFTER THYROID MEDICATION

# A THESIS

Submitted for Partial Fulfilment of Master Degree in Medicine

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( M. B., B. Ch.)



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AIM OF WORK

### AIM OF THE WORK

The mitochondrium is mainly responsible for the energy production in the cell, and thyroxine has a direct action on the mitochondrial function. 60% of the cellular magnesium is found in the mitochondrial component of the cells. Williams (1974).

Hypomagnesaemia can lead to muscle tremors, twitches and bizarre movements, a picture similar to hyperthyroidism, while hypermagnesaemia produces muscle weakness and somnolence which simulates the picture of hypothyroidism. Flink et al. (1957).

Copper has been recognized as an essential dietary element. It is necessary for promotion of homeostasis. Elvehiem (1935).

Several proteins and enzymes require copper as part of their molecular structure. These copper proteins include e.g. cytochrome oxidase, ceruloplasmin, tyreeinase, as well as histaminase. Prasad (1976).

Zine is acomponent of a number of enzymes particularly carbonic anhydrase, carboxypeptidase A and the dehydrogenases. It is also important in man for growth and sexual development, and it seems important for the synthesis of RNA and DNA, and for cell division. Varley et al. (1980).

The aim of this work is to study magnesium, copper, and zinc metabolism inhypothyroid cases, and the changes which occur in serum and urinary magnesium, copper and zinc before and 2 weeks after thyroxine therapy, and to demonstrate the role of these changes in hypothyroid manifestations.

REVIEW OF LITERATURE

MAGNESIUM METABOLISM

# Normal Distribution and Turnover of Magnesium in Man Magnesium body contents:

The total body content of magnesium is 200 m.mols (in a 70 kgm.man). It is about 21 gm of magnesium.

About 70 % is combined with calcium and phosphorus in the complex salts of bones, the remainder is in the soft tissues and body fluids. Harper (1973).

The range obtained by atomic abrorption spectrophotometry for whole blood magnesium is 1.4 - 2.5 m eq./liter (average 2.0) (1.7 - 3.0 mg/100 ml). Lanter (1975). The levels of magnesium in serum of healthy people are remarkably constant, remaining on the average at 1.7 m.eq./liter, and varying less than 15 % from this mean value. Wacker and Parisi (1968).

Serum magnesium measures 0.7 - 0.95 m.mol./liter (1.8-2.3 mg/100 ml.). Price (1978).

About 80 % of serum magnesium is in a diffusible state, the nondiffusible fraction being probably combined with serum proteins. Lanter (1975).

In the cerebrospinal fluid, magnesium level is higher than that of blood (2.5 - 3.0 m.eq/liter) (3.0 - 3.6 mg/100 ml). Lanter (1975).

According to Fomon (1974) human breast milk contains about 4 mg/100 ml.

### Requirements and intake :

The daily requirement of magnesium in the diet is 350 mg/day for adult men and 300 mg/day for adult women. Harper (1973).

For the neoborn it is much lower, being 60 mg/day, and increases to 150 mg/day for children. Fomon (1974)

### Biochemistry of Magnesium

Atomic composition of magnesium makes it able to capture and transport energy effeciently. Also, magnesium atom can hold reacting groups effeciently together. Magnesium atom has a chelating function which is necessary for all the photosynthetic processes. Prasad (1974).

Magnesium will prove to be another divalent cation with a messenger function as that of calcium in the mediation of hormonal functions. Rasmussen (1974).

### Magnesium Metabolism

### Absorption:

Ingested magnesium is mainly absorbed by the small intestine particularly the ileum, and the average daily absorption of magnesium from the gastrointestinal tract is 0.14 m.eq/kg. Fletcher et al. (1960).

There is an interrelationship between absorption of magnesium and calcium, as there is a common transport

mechanism for both cations across the intestinal Wall. Alcock and MacIntyre (1962).

The main factor controlling the amount of magnesium absorbed in a given time is the ionic concentration of magnesium in the digested food at the absorption
site, apart from the small effect of the potential difference across the wall of the small intestine. High
intake of phosphate, calcium and alkalies appears to
diminish magnesium absorption, while high magnesium
intake increases urinary calcium. Smith and Mc Allan
(1966).

# Secretion of magnesium:

Magnesium is secreted through the liver and pancreas into the bile and pancreatic juice which is followed by complete reabsorption. This explains hypomagnesaemia in patiants loosing high amounts of intestinal fluids. Lear and Gron (1968).

### Urinary excretion of magnesium:

The urinary excretion of magnesium has diurnal variation, being lowest during the normal sleeping hours and **-** 7 **-**

maximal in the late morning. Lanter (1975).

Most of absorbed magnesium is excreted by the kidney, and fecal magnesium represents largely the unabsorbed fraction. Aikawa et al. (1960).

Magnesium filtered by the glomeruli is reabsorbed by the renal tubules by an active process, and magnesium may be secreted by the renal tubules. It is noted that mercurial and thiazide diuretics as well as acidifying substances increase the urinary excretion of magnesium and calcium. Prasad (1976).

Magnesium is also excreted in sweat which can acount for 25 % of magnesium lost daily. Consolazio et al.(1973).

### Magnesium Homeostasis

It is the ionic fraction of magnesium which appears in the glomerular filtrate, while protein bound magnesium filtered is returned to the circulation through lymph. So, hypomagnessemia can result from renal diseases with heavy proteinuria. There is always maximal tubular absorption of magnesium. So, magnesium defeciency does not occur even under low magnesium diets. Barnes et al. (1958).