APPLICATION OF ENVIRONMENTAL ISOTOPES

TO THE STUDY OF WATER AND SALINITY BUDGETS OF

THE IRRIGATION NETWORK OF THE NILE DELTA

Submitted by

MOSTAFA AHMED AWAD ALY EL-GAYYAR

M.Sc. (Chemistry)

Nuclear Chemistry Department

Nuclear Research Centre,

Atomic Energy Establishment

For

The Degree of Doctor Of Philosophy of Sciences
(in Inorganic Chemistry)

То

546.24 MA

Chemistry Department
Faculty of Science
Ain Shams University
Cairo, Egypt

27796

1988

## APPLICATION OF ENVIRONMENTAL ISOTOPES TO THE STUDY OF WATER AND SALINITY BUDGETS OF THE IRRIGATION NETWORK OF THE NILE DELTA

Thesis Adivsors:

Prof. Dr. Said Thabit Abd El-Rehim

Prof. Dr. M. S. Hamza

Prof. Dr. F. M. Swailem

Ass. Prof. Dr. A. I. M. Aly

Approved

M. Hamaza F. Swailem

A.I.M. Aly

Prof. Dr. N.M. Guindy

Head of Chemistry Department, Faculty of Science -Ain Shams University.



## ACKNOWLEDGEMENTS

The Candidate is greatly indebted to Prof. Dr. S.Thabit Abd El-Rehim, Prof. of Physical Chemistry, Faculty of Science, Ain Shams University for his supervision of this work.

The Candidate expresses his gratitude to Prof.

Dr. M.S.Hamza, Nuclear Research Center, Egyptian Atomic

Energy Authority, Prof. Dr. F.M. Swailem, Middle Eastern

Regional Radioisotope Center for the Arab Countries, Dokki,

Cairo (M.E.R.R.C) and Dr. A.I.M.Aly, Associate prof., Nuclear

Research Center, Atomic Energy Authority, for their supervision, guidance and encouragement through the course of

this research and valuable suggestions and discussuions.

The successful completion of my training at

Lamont Geochemistry Department of Columbia University, N.Y.,

U.S.A. is due to the encouragement of Prof. Dr.H.James

Simpson.

The author wishes to extend his thanks to Prof.

Dr. Abdel - Gawad Sayed Emara, Head of Nuclear Chemistry Department, Nuclear Research Center, Atomic Energy Authority

for his kind support and encouragement.

9

The author wishes to thank also all the colleagues of the Nuclear Chemistry Department for their kind cooperation. Thanks are also due to Chemical Engineer, Wafaa. M. Moustafa, who did a superb job in drafting the figures.

Thanks to my family for their encouragement during the whole work .

## CONTENTS

		Pag
Acknow	ledgement	i
	CHAPTER 1	
	GENERAL INTRODUCTION	
7 7		
1.1.	Introduction	1
1.2.	Reuse of Drainage Water	6
1.3.	Sampling Program	8
	AIM OF WORK	9
	CHAPTER 2	
	EVAPORATION RATES ESTIMATED BY	
	STABLE ISOTOPES	
2.1.	Introduction	10
2.2.	Enrichment of Heavy Isotopes during Evapo-	
	ration	12
2.3.	Estimation of Heavy Isotope Enrichment	
	Trends from Evaporation Pan Experiments	14
2.4.	Methodology	16
2.5.	Results and Discussion	16
2.6.	Water Budget	32
2.6.1.	Irrigation water input	34
2.6.2.	Precinitation	7.0

		Pag
2.6.3.	Seepage to groundwater aquifers	36
2.6.4.	Drainage water transferred to the Sea	38
2.6.5.	Estimation of evapotranspiration	42
2.7.	Conclusions	46
	CHAPTER 3	
S	OURCES OF SALINITY IN THE DELTA AGRICULTURAL	
	DRAINAGE WATER	
3.1.	Introduction	48
3.2.	The Origin and Significance of the Major	
	Elements in Natural Water	49
3.3.	Sampling	60
3.4.	Drainage Systems	60
3.4.1.	Disposal of Drainage water	62
3.4.2.	The measuring network points	64
3.5.	Results and Discussions	65
3.6.	Sources of Salinity and Calculation Proce-	
	dures	97
3.7.	Mass Balance Approach	104
3.8.	Validation of the Method of Salt Balance	
	Calculation	143
<b>3.9.</b>	Minor Constituents in Agricultural Drainage	
	Water	150
3.10.	Conclusions	151

	Page
CHAPTER 4	
GREATER CAIRO WATER SUPPLY	
4.1. Introduction	154
4.2. Tritium in Surface Water of Egypt	
4.3. Greater Cairo Water Supply	
4.3.1. General description	
4.4. Cairo Area Aquifer Stratigraphy	
4.5. Mostorod and El-Maadi Well Field Samp	
4.6. Analytical Results	•
4.7. Conclusions	
APPENDICES	
ANALYTICAL TECHNIQUES	
Appendix (1) Determination of Silicate	••••• 190
Appendix (2) pH and Alkalinity	
Appendix (3) Conductivity	_
Appendix (4) Atomic Absorption Spectrometry	2,0
Measurements for Determination	of
Cations	
Appendix (5) Ion Chromatography with Conduct	
Measurements for Determination	_
Anions	•

		Page
	Oxygen-18 and Deuterium Measurements Tritium Enrichment and Counting	213
	Measurements	226
SUMMARY	•••••••••••	250
REFERENCES	•••••••••••	256
Arabic Summary.		

CHAPTER 1.

GENERAL INTRODUCTION

## I.1. Introduction:

All forms of life on earth depend on water. Humans use water for many purposes, drinking, irrigation fisheries, industrial processes, transporation and waste disposal. In recent years however, the enormous impacts of humans on natural systems have been recognized. It has been further established that any program to regulate human activities involving water must be based on an understanding of natural processes affecting the water cycle.

In many areas of the world especially in the arid zones, water is and has been a scarce commodity. World water resources are being increasingly strained to meet requirements from the growth of population, increasing industrialization and intensified, agriculture. The exploitation of more water resources is essential and therefore great efforts have to be made to develop the available water resources.

The Nile is usually considered the longest river in the world. It represents the only major renewable source of water in an extremely arid zone. More water for irrgiation became available with the building of the old Aswan Dam in 1902. When enlarged in 1912 and again 1933, it reached a total capacity of  $6 \times 10^9$  m<sup>3</sup> of water. The construction of the Aswan High Dam (1960-1967) eliminated the Nile's seasonal floods and allowed all agricultural land to be brought

under pernnial irrigation.

Egypt has a total land area of about one million  $Km^2$ . It belongs to arid and semi arid regions, of which  $3.75 \times 10^6$  Feddans of Agricultural land are concentrated in the Delta (Ministry of Irrigation 1981). The water of the River Nile is not sufficient for the purposes of significant expansion of the cultivated areas outside of the Nile Delta and also valley and increasing the existing cropping intensity. Thus, agricultrual development plans depend mainly upon better use of existing water sources, and development of new water resources. One of the main technical problems in water to optimite crop growth is, therefore, to establish of the quantity of water to be applied for irrigation of a particular area and the quantity of drainage water and salinity that may be removed from the soil in order to meet the optimal growth situation of the crops.

About half of Egypt's population is involved in agricultural. The silty clay layers used for crop growth average 10 m in depth and overlie thicker layers of sand and gravel. Lighter textured soils occur along the fringes of the Delta and in the Nile Valley.

At present approximately 2-3 billion m<sup>3</sup> of drainnage water is being reused for irrigation purposes yearly.

Land development projects are planned that will require

another 4 billion m<sup>3</sup> of drainage water in order to supply part of envisaged extension of the agricultural acreage in the future.

Under the present conditions about 15% of drainage waters with suitable quality is reused (Rijtema and Roest 1982) but an enormous quantity of potentially useable drainage water is flowing to the sea. A portion of this water that has a low salinity could be reused or saved making irrigation water available for additional areas to be reclaimed.

The ultimate goal with respect to the use of the drainage water resource is 10 billion m<sup>3</sup> yearly to be reached by 1990.

The acceptance of the reuse of drainage water as an official policy in water management requires a careful and detailed understanding of the processes controlling the salinity of drainage water, groundwaters and the quantities of water represented by each of the major loss terms:

- 1. Evaporation
- 2. Transpiration
- 3. Seepage.

Variations in the stable isotope composition of water (ratios of heavy isotopes to light isotopes  $0^{18}/0^{16}$  and  $H^2/H^1$ ) are extremely sensitive to evaporation in drainage water and these variations could provide a potentially

precise indicator of the total amount of evaporation which has occurred. Since transpiration produces very little change in the stable isotopic composition of water paning through irrigated fields. Gaging information combined with the stable isotope data can indicate directly the efficiency of water use in the entire irrigated crop system.

Calibration of the stable isotopic composition data to the amount of evaporation loss can be accomplished by sequantital sampling of water from an open pan filled with irrigation water and allowed to evaporate to small fraction of initial volume. Thus samples from experimental evaporation pan and from drainage water can be provide new information on the efficiency of irrigated water use.

The stable isotope data combined with chemical data can be used to indicate the sources of salinity in the drainage waters. These data can help in resolving the relative contribution of salts from:

- 1. evaporative concentration,
- 2. soil leaching,
- 3. possible upward mixing of marine salts in areas with potential input from intrusion of sea water into deeper groundwaters.

Salinity increases derived from soil leaching should produce no change of the isotope composition of drainage water, while those resulting from either evaporation concentration or mixing of marine water would be

accompanied by large stable isotopic changes. The changes of stable isotope composition induced by these latter two processes are very different and should be possible to be largely resolved in combination with chemical data.

Together with hydrological and hydrochemical investigations of water resources, radiometric investigation are also important. The determination of tritium content present in undergroundwaters is a useful tool for the determination of the age of recharge of these waters. It gives an idea about whether groundwater are "old" or rapidly recharged from modern sources. Applying tritium dating technique, the recharge rate of groundwater bodies can be computed in selected areas. This tracer has been present in high concentrations in irrigation water only since the mid 1950's with the highest concentrations present in the middle 1960's. Thus groundwaters from the Nile Delta having amount of this isotope comparable to those present during peak years, have been recharged to a significant extent from the irrigation network and annual flooding of the Nile River between the late 1950's and the late 1960's while groundwater free of tritium have not been recharged within the last 30 years.

The combination of evaporation pan stable isotopes data and chemical data in drainage waters from the Nile Delta and tritium data from selected groundwaters, should provide important new information for management of the