

**Ain Shams University University college for Women** (Arts, Science & Education)

### SYNTHESIS, CHARACTERIZATION AND POLYMERIZATION OF A NOVEL ACRYLAMIDE

A Thesis Submitted in Partial Fulfillment of the Requirements for the Degree of

### DOCTOR OF PHILOSOPHY in PHYSICAL CHEMISTRY

by

### **Fatma Ahmed Mahmoud Gomaa**

To

Department of Chemistry University college for Women (Arts, Science & Education) Ain Shams University

#### **UNDER SUPERVISION OF:**

### Prof. Dr. Samia M. Mokhtar

Professor of Physical Chemistry University College for Women Ain Shams University, Egypt

#### Prof. Dr. Maher Z. Elsabee

**Professor of Physical Chemistry** Faculty of Science,

Cairo University, Egypt

#### Dr. Sakina M. Abd-ElAziz

Lecturer of Physical Chemistry University College for Women Ain Shams University, Egypt

Ain Shams University 2014



### APPROVAL SHEET

# SYNTHESIS, CHARACTERIZATION AND POLYMERIZATION OF A NOVEL ACRYLAMIDE

A Thesis Submitted in Partial Fulfillment of the Requirements for the Degree of

# DOCTOR OF PHILOSOPHY in PHYSICAL CHEMISTRY

by

### **Fatma Ahmed Mahmoud Gomaa**

Under Supervision of:

#### Prof. Dr. Samia M. Mokhtar

Professor of Physical Chemistry University College for Women Ain Shams University, Egypt

#### Prof. Dr. Maher Z. Elsabee

Professor of Physical Chemistry Faculty of Science, Cairo University, Egypt

#### Dr. Sakina M. Abd-ElAziz

Lecturer of Physical Chemistry University College for Women Ain Shams University, Egypt

**Head of Chemistry Department** 



### **QUALIFICATION**

Student Name : Fatma Ahmed Mahmoud Gomaa

Scientific Degree : M.Sc. (Chemistry)

Department : Chemistry

Faculty : University College for Women

(Arts, Science and Education)

University : Ain Shams

B.Sc. Graduation Date : May 2005

M. Sc. Graduation Date : October 2010

Ph.D. Graduation date : 2014







### **ABSTRACT**

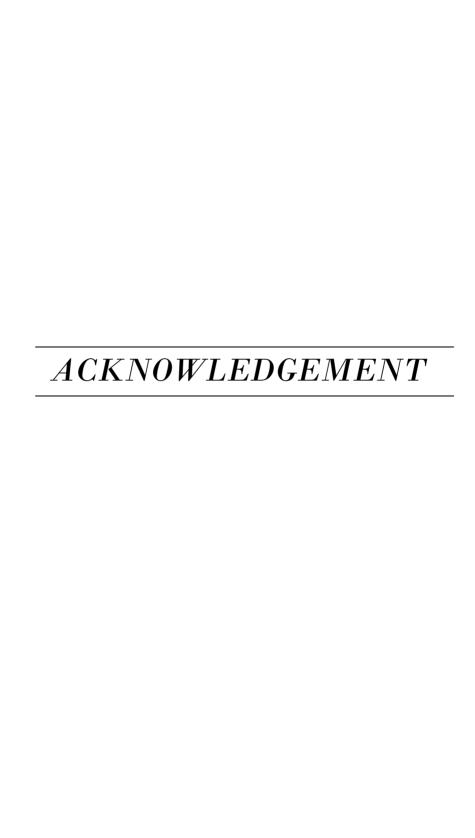
## SYNTHESIS, CHARACTERIZATION AND POLYMERIZATION OF NOVEL ACRYLAMIDE

#### Student Name: Fatma Ahmed Mahmoud Gomaa

University College for Women (Arts, Science and Education), Ain Shams University

Two novel acrylamide monomers 4-(4'-cyanophenoxy) phenyl acrylamide (CPPAM) and 4-(4`-trifluoromethyl) phenoxy N-phenyl acrylamide (FPPAM) were synthesized and characterized by elemental analysis. The free radical- initiated polymerization of CPPAM and FPPAM were carried out in THF and 1.4 dioxane respectively, using azobisisobutyronitrile as initiator. The two monomers and their polymers were investigated by <sup>13</sup>CNMR, <sup>1</sup>HNMR and FTIR. The effect of monomer, initiator concentrations and temperature on the rate of polymerization (R<sub>n</sub>) was studied. The molecular weight averages (M w and M n) and polydispersity index of the poly-FPPAM were determined by gel permeation chromatography and were equal to 411,061, 129,046 and 3.185 respectively. The properties of Poly CPPAM and Poly FPPAM including thermal behavior, thermal stability, photo-stability, solubility and intrinsic viscosity were studied. Moreover, the monomer reactivity ratios for the copolymerization of CPPAM and FPPAM with MMA, VAc and AN were calculated by three different methods. It has been found that the PCPPAM, PFPPAM and the copolymers have high thermal stability.

**Keywords:** Fluoropolymers, cyanopolymers, acrylamide, FTIR, NMR, Free-radical polymerization, Copolymerization kinetics, Thermal analysis, Methyl methacrylate, vinyl acetate and acrylonitrile.



### ACKNOWLEDGEMENT

I am deeply thankful to **Allah**, by the grace of whom the progress and success of this work was possible.

It is my great pleasure to cordially express my sincere appreciation of all the efforts provided by Prof. Dr. Samia M. Mokhtar, Professor of Physical Chemistry, and Ex. Head of Chemistry Department, University College for Women (Arts, Science and Education), Ain Shams University. I am deeply indebted to you for suggesting the subject and close supervision. You have always had time to listen, discuss and comment on my ideas and manuscripts. I especially appreciate all the time you spent on my work and me during the last intensive months.

I am deeply indebted to **Prof. Dr. Maher Zaki ElSabee** Professor of Physical Chemistry, Faculty of Science, Cairo University for his continuous support, encouragement, fruitful opinions, kind supervision and help during all the phases of achieving the objectives of this thesis.

I would like to express my deepest thanks and gratitude to my supervisor Dr. Sakina M. Abdel Aziz; Lecturer of Chemistry; Chemistry Department, University College for Women (Arts, Science and Education), Ain Shams University for her kind supervision, guidance, faithful help, and support during the whole process.

I would like to express my great thanks to all members of the Chemistry Department, University College for Women (Arts, Science and Education, Ain Shams University for their continuous helping and support.

Very special thanks to my father, mother, husband, kids and my brothers and sisters for continuous care, support and encouragement.

Fatma Ahmed Mahmoud Gomaa Cairo, May 2014



### List of Contents

List of Tables	viii
List of Figures	xi
List of Symbols and abbreviations	xviii
Aim of the Work	XX
Chapter One	
INTRODUCTION & LIT SURVEY	ERATURE
1.1 Introduction	1
1.2 Fluoropolymers	4
1.3 Acrylic Monomers	
1.3.1 Acrylic Monomers Synthesis	
1.3.2 Polymerization of Acryloylamido M	(onomers 20
1.3.3 Copolymerization of Acryloyl Amid	o 24
1.3.4 Thermal Stability	30
Chapter Two	
Materials and Experimenta	al Techniques
2.1 Materials	34
2.1.1 Monomers	34
2.1.1.1 Preparation of 4-4`-Aminopheno Acrylamide	•
2.1.1.2 Preparation 4-(4`-trifluorom	nethyl) phenoxy

2.1.1.3 Methyl Methacrylate	
2.1.1.4 Vinyl Acetate	
2.1.1.5 Acrylonitrile	
2.1.2 Initiator	
2.1.3 Solvents	•
2.2 Experimental Techniques	
2.2.1 Polymerization Procedure	
2.2.2 Kinetics of Homopolymerization	
2.2.3 Viscosity Measurements	
2.2.3 Gel Permeation Chromatography (GPC)	
2.2.4 Thermogravimetric Analysis (TGA)	
2.2.5 Differential Scanning Calorimetric Analysis	
2.2.6 Photo Stability	
2.2.7 Solubility	
2.2.8 X-ray Diffraction	
2.3 Copolymerization	
2.3.1 Copolymerization Process	
2.3.2 Copolymer Analysis	
2.3.3 Calculation of the Reactivity Ratios of the monomers	
2.3.3.1 Fineman-Ross (F-R) method	
2.3.3.2 Kelen-Tudos (K-T) method	
2.3.3.3 Non-Linear Least Square Method (NLLS)	
234 Thormal Analysis	

### **Chapter Three**

### **Results and Discussion**

### **Part One**

Monomers	<b>Synthesis</b>	, Po	lymerizatio	n and
	Characte	eriza	ation	

<b>3.1</b> Mon	nomer Synthesis
<b>3.2 Hon</b>	nopolymerization of CPPAM and FPPAM
3.3 Poly	merization Kinetics of CPPAM and FPPAM
3.3.1	Dependence of the Rate of Polymerization on Initiator Concentration
3.3.2	Dependence of the Rate of Polymerization on the Monomer Concentration
3.3.3	Effect of Temperature on the Rate of Polymerization
<b>3.4 Cha</b>	racterization of the Prepared Polymers
3.4.1	Viscosity Measurements
3.4.2	Gel Permeation Chromatography (GPC)
3.4.3	X-Ray Diffraction (XRD)
3.4.4	Solubility
3.4.5	Photo Stability of Poly-CPPAM and Poly-FPPAM
3.4.6	Thermal Stability
	3.4.6.1 Thermal Analysis of PCPPAM
	3.4.6.2 Thermal Analysis of PFPPAM

### **Part Two**

# Copolymerization of the New Synthesized

### **Monomers**

-	olymerization Behavior of Both CPPAM and PAM	89
	Copolymerization Behavior of 4-(4`- CyanoPhenoxy) Phenyl Acrylamide Monomer (CPPAM)	89
3.5.2	Copolymerization Behavior of 4-(4`- Trifluoromethyl) Phenoxy N-Phenyl Acrylamide Monomer (FPPAM)	94
3.6 Chai	racterization of Copolymerization	99
3.6.1	Spectral Characterization of Copolymer	100
3.6.2	Thermal Behavior of Copolymers	103
SUMMA	ARY AND GENERAL CONCLUSIONS	113
Referen	ces	115
Arabic S	Summary	128