

# 127, 17 27, 17 (20) 77, 17 (20









## جامعة عين شمس

التوثيق الالكتروني والميكروفيلم



نقسم بللله العظيم أن المادة التي تم توثيقها وتسجيلها علي هذه الأفلام قد اعدت دون آية تغيرات



### يجب أن

تحفظ هذه الأفلام بعيداً عن الغبار

في درجة حرارة من 15-20 مئوية ورطوبة نسبية من 20-40 %

To be kept away from dust in dry cool place of 15 – 25c and relative humidity 20-40 %



ثبكة المعلومات الجامعية





Information Netw. " Shams Children Sha شبكة المعلومات الجامعية @ ASUNET بالرسالة صفحات لم ترد بالأص



#### The Effect of Nile Pollutants From Toxic Heavy Metals and Natural Radioelements Wastes on The Physiology of Certain Aquatic Weeds

By
Yasser Sayed Mahmoud Ramadan
B. Sc. and M.Sc. (Botany)

Thesis
Submitted fot the Degree of
Doctor of Philosophy
IN ( Botany)

Hasson Fode

Supervised By

Prof. Dr. Hassan A. Foda

Piùi. Di nassali A. Fuua

Professor of Plant Physiology
Ain Shams University

Prof.Dr.Mohamed B. Hafez

Professor of Radiation Chemistry

Hot Laboratories Center.

Atomic Energy Authority

Raifa A. Hassanei

Prof. Dr. Raifa A. Hassanein

Professor of Plant Physiology

Ain Shams University

N. Hafe>

Prof. Dr. Nabil S. Hafez

Professor of Radiation Chemistry

Hot Laboratories Center

Atomic Energy Authority

**Botany Department** 

Faculty of Science

Ain Shams University

1999

2012 de 1

ع. د جور مراز عام المرجة بتاريخ ع. د جور مراز عام عام المرجة بتاريخ على المرجة بتاريخ عد ريم المرد على منع المرجة بتاريخ

Ž

#### Approval Sheet

Title of Thesis: The Effect of Nile Pollutants From Toxic Heavy Metals and Natural Radioelements Wastes on The Physiology of Certain Aquatic Weeds

By
Yasser Sayed Mahmoud Ramadan
(B. Sc. & M. Sc.)

This Thesis for Ph. D. Degree has approved by:

#### Prof. Dr. Raifa Ahmed Hassanein.

Professor of Plant Physiology, Faculty of Science, Ain Shams University.

#### Prof. Dr. Mohamed Bader Hafez.

Professor of Radiation Chemistry, Hot Laboratories center, Atomic Energy Authority.

#### Prof. Dr. Taymour Hussain Kamal.

Professor of Radiation Biology, Nuclear Research Center, Atomic Energy Authority.

#### Prof. Dr. Mahmoud Mohamed Ali Gaber.

Professor of Plant Physiology, Faculty of Science, Cairo University.

Date: / / 1999

#### **ACKNOWLEDGMENT**

This work has been carried out under the supervision of Prof. Dr. Hassan Anwar Foda. Professor of plant physiology Faculty of Science, Ain Shams University, Prof. Dr. Raifa Ahmed Hassanein, Professor of Plant Physiology Faculty Science, Ain Shams University, Prof. Dr. Mohamed Bader Hafez. Professor of Radiation Chemistry, Hot Laboratories Center, Atomic Energy Authority, and Prof. Dr. Nabil El-Sayed Hafez, Head of Radiation Protection Department, Atomic Energy Authority. The author wishes to express the deepest thanks and gratitude to them for topic, continuous supervision, suggesting the sincere guidance, useful discussions and help in the final accomplishment of this work.

Thanks for all staff members in Radiation Protection Department, Atomic Energy Authority for sincere help

Thanks are also to the Head of the Botany Department and all staff members in the same department, Faculty of Science, Ain Shams University for sincere help and fruitful advises.

Thanks are also to my family for continuous encouragement.

.

#### **ABSTRACT**

Name: Yasser Sayed Mahmoud Ramadan

**Title**: The Effect of Nile Pollutants From Toxic Heavy Metals and Natural Radioelements Wastes on The Physiology of Certain Aquatic Weeds

Degree: Doctor of Philosophy

Submitted to: Botany Department, Faculty of Science,

Ain Shams University.

Due to large production of both low level radioactive liquid and toxic industrial wastes, aquatic plants are used to absorb undesirable ions from waste solutions. The present work illustrates, the uptake of some metal ions existed in low level liquid radioactive or industrial wastes namely U VI, Th IV, Ce III, Fe III, KMnO4, Cu II. Pb II, Ni II, Cd II and phenol by Eichornia crassipes or Potamogeton crispus plant. The uptake process was performed for each metal ions, and in presence of some additives then for groups of metals, which represent, the radioactive and toxic wastes. The changes photosynthetic pigments of the leaf, toxicity index and the carbohydrate contents of Eichornia crassipes and Potamogeton crispus plants before, during and after the uptake process were determined. The obtained results were supported bγ electron microscope applying techniques to examine the photosynthetic apparatus.

The effects of some environmental parameters such as pH, temperature, concentration, interference of some additives and plant age on the uptake and accumulation of the studied elements by the two aquatic plants were also investigated.

The variation in mineral composition of *Eichornia* crassipes plant at different sites along Ismailia canal compared with the mineral composition in water at these

sites showed that, Fe, Ni, Cd and Pb ions were accumulated in roots in greater amounts than in shoots, while a reverse pattern was shown concerning Cu ions accumulation. In addition, the distribution of Mn ions was shown to be nearly the same in both shoots and roots of *Eichornia crassipes* plants. It is of interest to mention here that one gram of *Eichornia* roots can accumulate 8.2, 7.2, 3.4, 1.6 and 1.2 times from Fe, Mn, Cd, Ni and Pb ions respectively as those recorded in water. These absorbed metal ions did not affect the photosynthetic pigment contents and the carbohydrate levels indicating that this plant is tolerant to the absorbed concentrations of metal ions.

The two examined plants revealed specific pattern for the uptake of studied metal ions, with regard to time. They could be used as bioindicator to monitor their existence in fresh water environment. They can detect such metal ions if possible radioactive contamination in the surrounding environmental water is occurred. Eichornia crassipes and Potamogeton crispus could be used eventually for treatment of low level liquid radioactive or toxic wastes. It can be recommended that the treated plants with the studied metal ions should be collected directly after 10 hours of the beginning of the uptake.

Contents	Page
List of Figures	
List of Tables	, i
List of Plates	
Introduction	1
Review of Literature	4-29
Material and Methods	30-40
A- Material	30
1- Experimental aquatic plants	30
2- Chemicals	31
3- Apparatus	33
B- Experimental Methods	35
1- Maintenance of plants	35
2- Determination of thorium	35
3- Determination of uranium	36
4- Determination of heavy metals in plant tissues	36
5- Determination of the photosynthetic pigments	37
6- Estimation of carbohydrates	37
7- Transimission electron microscopic determination	
Safety precautions	
Results	41
I - Effect of Toxic Ions on Certain Aquatic Plants	41
1- Effect of Toxic Ions on Eichornia Crassipes	
1.1- Uptake of toxic ions by Eichornia crassipes	41
1.1.1- Uptake of UVI by Eichornia crassipes	41
a- Effect of Eichornia crassipes age on the uptak process	e 41
b- Effect of pH of U VI solution on the uptake process by Eichornia crassipes	44
c- Effect of temperature on the uptake of U VI solution by Eichornia crassipes	46
d- Effect of concentration on the uptake of U VI by Eichornia crassipes	48
e- Uptake of U VI by Eichornia crassipes in presence of different additives	53

1- Uptake of UVI by Eichornia crassipes in	53
presence of NaCl	
2- Uptake of UVI by Eichornia crassipes in	56
presence of MgCl <sub>2</sub>	
3- Uptake of UVI by Eichornia crassipes in	59
presence of CaCl <sub>2</sub>	
4- Uptake of UVI by Eichornia crassipes in	59
presence of FeCl <sub>3</sub>	
5- Uptake of UVI by Eichornia crassipes in	62
presence of KNO <sub>3</sub>	
6- Uptake of UVI by Eichornia crassipes in	68
presence of $(NH_4)_2SO_4$	
7- Uptake of UVI by Eichornia crassipes in	68
presence of $Pb(NO_3)_2$	
1.1.2- Uptake of Th IV by Eichornia crassipes	72
1.1.3- Uptake of Ce III sulphate by Eichornia crassines	72
1.1.4- Uptake of Fe III by Eichornia crassines	78
1.1.5- Uptake of KMnO <sub>4</sub> by Eichornia crassines	82
1.1.6- Uptake of Cu II by Eichornia crassines	82
1.1.7- Uptake of Pb II by Eichornia crassines	85
1.1.8- Uptake of Cd II by Eichornia crassipes	90
1.1.9- Uptake of Ni II by Eichornia crassipes	93
1.1.10- Uptake of phenol by Eichornia crassines	93
1.1.11- Uptake of U VI, Th IV, Fe III and KMnO <sub>4</sub> from	98
simulated industrial wastes by Eichornia crassing	5
1.1.12- Uptake of mixture from Pb II, Cd II and Ni II	98
from simulated industrial wastes by Eichornia	
: crassipes	
1.1.13- Uptake of phenol from industrial wastes	100
(coal tar) by Eichornia crassipes	•
1.2- Changes in photosynthetic pigments content of	104
Eichornia crassipes leaves in response to the	
uptake of metal ions	
1.3- Changes in carbohydrates contents of Eichornia	123
crassipes in response to the uptake of metal ions	
1.4- Ultrastructural Changes in the Eichornia cells due	132
to their immersion in metal ion solutions	

.

\*

.