

PREPARATION AND CHARACTERIZATION OF SOME ANTIBACTERIAL GLASSES

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By

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تحضير وتوصيف بعض أنواع من الزجاج المضاد للبكتيريا

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كجزء متمم للحصول على درجة ماجستير العلوم في الكيمياء

من

أحمد محمد أحمد الفقى

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ABSTRACT

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Abstract

This work was carried out to investigate the antibacterial effects of some undoped and silver-doped P₂O₅ - CaO - Na₂O glasses against S.aureus, P.aeruginosa and E.coli microorganisms using agar disk-diffusion assays. Glass forming regions in the ternary P₂O₅ - CaO -Na₂O and in the quaternary P₂O₅ - CaO - Na₂O - Ag₂O systems were determined. Density, molar volume, dissolution of glass in water, pH changes of water during glass dissolution, and concentrations of silver ions released from silver-doped glasses into water during their dissolution were determined. The structures of some glasses were studied by XRD, FT-IR, and UV-VIS spectroscopy. The tested silver-free and silver-doped glasses demonstrated different antibacterial effects against the tested micro-organisms. For silver-free glasses, an increase in inhibition zone diameter (zone of no bacterial growth) was seen with the increase in the glass dissolution rate and with the decrease in water pH. Silver-doped glasses showed an increase in inhibition zone diameter with increasing Ag₂O content. An increase in concentrations of silver ions released from silver-doped glasses into water was seen with increasing time of glass dissolution and with increasing Ag₂O content. The dissolution rates of P₂O₅ - CaO - Na₂O glasses decreased with increasing CaO content and slightly decreased with gradual replacement of Na₂O by Ag₂O.

Keywords: Antibacterial glasses; Silver-doped phosphate-based glasses; Glass dissolution; Controlled release; Silver ions; Antibacterial effect.

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ملخص البحث:

تختص هذه الرسالة بدراسة التأثير المضاد للبكتيريا لبعض التراكيب الزجاجية في النظام - P2O5 - CaO - Na2O Ag₂O و المحتوية على ٠ و ٥ . • و ١ و ٢ مول % من Ag₂O وذلك على بعض أنواع من البكتيريا مثل S.aureus و P.aeruginosa و E.coli با ستخدام طريقة (Agar disk-diffusion). وقد تم تحضير العديد من التراكيب الزجاجية في النظام الثلاثي P2O5 - CaO - Na2O - Ag2O و في النظام الرباعي P2O5 - CaO - Na2O - Ag2O باستخدام طريقة الصهر التقليدية وذلك بهدف تحديد مدى التراكيب الكيميائية في هذه الأنظمة والتي تعطى عند صهرها زجاجا. تمت دراسة بعض الخواص مثل الكثافة و الحجم المولاري و ذوبان الزجاج في الماء و تغير الرقم الهيدروجيني للماء أثناء ذوبان الزجاج بالإضافة الى تقدير تركيزات أيونات الفضة الذائبة في المحاليل المائية. كما تم دراسة التركيب البنائي لبعض التراكيب الزجاجية وذلك باستخدام حيود الأشعة السينية و إمتصاص الأشعة تحت الحمراء و الأشعة المرئية و فوق البنفسجية. وقد بينت نتائج إختبارات المزارع البكتيرية أن جميع التراكيب الزجاجية المختبرة (المحتوية و غير المحتوية علىAg2O) لها تأثيرات مضادة للبكتيريا وذلك بدرجات متفاوتة تبعا لنوع البكتيريا (S.aureus > P.aeruginosa > E.coli) و تركيب الزجاج و تركيز Ag₂O به. و قد لوحظ أن التأثير المضاد للبكتيريا للتراكيب الزجاجية غير المحتوية على Ag₂O يزيد مع زيادة معدل ذوبان الزجاج وأيضا مع زيادة النقص في الرقم الهيدروجيني. وبالنسبة للتراكيب الزجاجية المحتوية على Ag2O فقد لوحظ زيادة التأثير المضاد للبكتيريا بزيادة تركيز Ag2O في الزجاج. كما تبين من دراسة ذوبان الزجاج في الماء أن معدل الذوبان للتراكيب الزجاجية في النظام الثلاثي P2O5 - CaO - Na2O يقل بزيادة تركيز CaO و أيضا يقل معدل الذوبان للتراكيب الزجاجية في النظام الرباعي P_2O_5 - CaO - Na_2O - Ag_2O وقد أوضحت قياسات الإمتصاص الذرى المحاليل الذوبان أن تركيز أيونات الفضة في المحلول يزيد بزيادة كلا من وقت ذوبان الزجاج وتركيز Ag2O في الزجاج. كما بينت دراسة إمتصاص أطياف الأشعة المرئية و فوق البنفسجية ظهور قمة إمتصاص عند طول موجي ٢٣٠ نانو متر تقريبا و التي تبين أنها ناتجة عن الإنتقالات الإلكترونية $4d^9 5s^1 - 4d^{10}$ لأيونات الفضة.

الكلمات الدالة: الزجاج المضاد للبكتيريا ؛ زجاج فوسفات الفضة ؛ أيونات الفضة ؛ التأثير المضادة للبكتيريا ؛ ذوبان الزجاج.

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List of symbols and abbreviations

W.H.O World Health Organization.

DNA Deoxyribonucleic Acid.

G + Gram-positive.

G - Gram-negative.

E.coli Escherichia coli.

P.aeruginosa Pseudomonas aeruginosa.

S.aureus Staphylococcus aureus.

MRSA Methicillin-Resistant Staphylococcus Aureus.

C. albicans Candida albicans.

ppm Part per million.

ppb Part per billion.

NASA National Aeronautic Space Administration.

ASTM American Society for Testing Materials.

PBGs Phosphate-Based Glasses.

CRGs Controlled Release Glasses.

 \mathbf{Q}^n PO₄ group with n = number of bridging oxygen atoms.

BO Bridging oxygen.

NBO Non-bridging oxygen.

NMR Nuclear Magnetic Resonance.

 T_{g} Glass transition temperature.

XRD X - Ray Diffraction.

FT-IR Fourier - Transform Infrared spectroscopy.

UV-VIS Ultraviolet - Visible spectroscopy.

Symbols and abbreviations

ICP-OES Inductively Coupled Plasma - Optical Emission Spectroscopy.

SBF Simulated Body Fluid.

SEM Scanning Electron Microscopy.

XPS X-ray Photoelectron Spectroscopy.

ESEM Environmental Scanning Electron Microscopy.

HBSS Hank's Buffered Saline Solution.

PGFs Phosphate-based glass fibres.

SHMP Sodium Hexa-Metaphosphate (50Na₂O - 50P₂O₅).

EDX Energy Dispersive X-ray.

MAS -NMR Magic Angle Spinning - Nuclear Magnetic Resonance.

XANES X-ray Absorption Near Edge Structure.

HEXRD High Energy X - Ray Diffraction.

CFUs Colony-Forming Units.

CDFF Constant depth film fermenter.

2D-NMR Two-dimensional NMR.

TEM Transmission Electron Microscopy.

C.difficile Clostridium difficile.

FAAS Flame Atomic Absorption Spectrometry.

Mass of glass sample in air.

M_L Mass of glass sample in liquid.

DIN Deutsches Institut für Normung (The German Institute for

Standardization)

ISO International Standards Organization.

MIC Minimum Inhibitory Concentration.

ATCC American type culture collection.

Symbols and abbreviations

GFR Glass-Forming Region.

 $\mathbf{D}_{\cdot \mathbf{R}}$ Dissolution Rate.

a.u. arbitrary units.

IZD Inhibition Zone Diameter.

T_m Melting temperature.

CLD Cross-Link Density.

δ Bending vibration mode.

 v_s Symmetric stretching vibration mode.

 v_{as} Asymmetric stretching vibration mode.

ATP Adenosine Tri-Phosphate.

RNA Ribonucleic Acid

NADH Nicotinamide Adenine Dinucleotide Hydrogenase.

APROVAL SHEET FOR SUBMISSION

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