



شبكة المعلومات الجامعية

بِسْمِ اللَّهِ الرَّحْمَنِ الرَّحِيمِ





شبكة المعلومات الجامعية



شبكة المعلومات الجامعية

التوثيق الالكتروني والميكرو فيلم

جامعة عين شمس

التوثيق الالكتروني والميكرو فيلم

قسم

نقسم بالله العظيم أن المادة التي تم توثيقها وتسجيلها
علي هذه الأفلام قد اعدت دون أية تغيرات



يجب أن

تحفظ هذه الأفلام بعيداً عن الغبار

في درجة حرارة من 15 – 20 مئوية ورطوبة نسبية من 20-40 %

To be kept away from dust in dry cool place of
15 – 25c and relative humidity 20-40 %



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بعض الوثائق الأصلية تالفة



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بالرسالة صفحات
لم ترد بالأصل



Ain Shams University
Faculty of Science

Phase Transition and Physical Properties of the Glassy System Ge-Se-Te

Thesis
Submitted for the Degree of
Doctor of Philosophy in
Physics

Submitted
Shadia Ali El-Said
M.Sc in Physics

Faculty of Science
Ain Shams University

Handwritten signature/initials.

1997

Handwritten signature/initials.

1992

إهداء

بوافر الود والعرفان اهدى هذا العمل لكل من :

الدكتور/ حسن رمضان وأولادى رانا وعمرو ووائل كما
أهديه لوالدتي وذكرى أبى.

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Abstract

Two binary systems $\text{Ge Se}_{2.5}$ and $\text{Ge Te}_{2.5}$ were first prepared, by direct fusion of elements, then were used to prepare the system $(\text{Ge Se}_{2.5})_{1-x} (\text{Ge Te}_{2.5})_x$ with $0 \leq x \leq 1$, by direct fusion under vacuum and quenching.

The vitreous state of resulting materials was confirmed by X-ray technique.

The density of the prepared glasses were determined by a hydrostatic method at room temperature. The data was fitted to a relation of the type

$$d = 4.223 + 1.67 \cdot 10^{-2} [\text{Te}], \text{ where Te is Te at. \% .}$$

The specific electrical conductivity σ of the vitreous materials was measured in the temperature range up to 150°C , below the softening points.

The activation energy of conduction E_σ was found to fit the form:

$$E_\sigma = -0.9788 - 17.743 \ln \sigma_{20}$$

DSC thermograms were obtained at three different heating rates (10, 15 and $20^\circ\text{C}/\text{min}$), for all glass systems. Three characteristic transitions could be identified which are due to softening T_g , crystallization T_p , and melting T_m processes.

DTA thermograms were obtained at different heating rates (5 to $40^\circ\text{C}/\text{min}$) for all samples. The activation energy of crystallization E_c were calculated using four different approaches:

i- Augis and Bennett, ii- Ozawa-Chen, iii- Takhor and iv- Kissinger methods.

The order of crystallization was obtained on basis of two different methods. Ozawa method and a method from the χ_p value.

To examine the state of crystallization the samples were left to crystallize at the highest T_p for around (7 hrs). X-ray diffraction studies were done on powdered samples.

The results were compared with those reported in literature

